Zoom HS Lesson Plan Template

Grade Level: High School

Approximate Timing: 90 Minutes

Skeleton Frame:

This lesson plan skeleton can be used to help organize different "Zoom Teacher Tools" throughout teachers' lessons to ensure the lesson is interactive, multimodal, and leverages the Zoom features in pedagogically effective ways to ensure students are engaged and learning. Teachers do not need to use a different "Zoom Teacher Tool" in each section but can look at them as building blocks that they can use in different combinations to enhance their instructional design by offering opportunities for students to connect, learn, and collaborate utilizing different tools and features.

Each section of the lesson plan includes one basic instructional component as well as options within each — since every day, lesson, and teacher is different!

NOTE: When designing instruction with additional supports (e.g., tutors, para professionals, special education teachers, residents, etc.) it can be helpful to increase and/or conduct small group breakout time in a rotational format to ensure students are getting the focused support they need. Additional adults can help monitor this time, lead small group instruction/discussions, as well as be an additional resource when students are working independently or as a whole group.

Lesson Plan Component	Zoom Teacher Tool Strategy	Approximate Timing
Welcome/Class Kick off:		10 minutes
Whole Group Instruction: Presentation Exploration Discussion		20-30 minutes
Small Group Break-outs:		30-40 minutes

Check for Understanding: Task/assignment Quick-check in Group/individual sharing Poll Assessment	10-20 minutes
Closing:	10 minutes

Example Lesson using the Zoom Teacher Toolkit:

High School Chemistry - Limiting & Excess Reagents

Lesson Plan Component	Zoom Teacher Tool Strategy	Approximate Timing
Welcome/Class Kick off: ● Fun entrance kick-off	 Welcome students and take attendance. "Virtual Non-verbal Cues" When students enter the Zoom room they are asked to put a reaction in their box that illustrates how they are feeling coming into class. This will help the teacher to identify students who may need a little more emotional support during the lesson. How are you feeling today?	5 minutes

Whole Group	Intro exploration: 5 minutes	30 minutes
Instruction:	S'more Stoichiometry Activity	
PresentationExploration	Activity Every student has a bag of	
Discussion	s'more ingredients that	
 Demonstration 	they received from their	
	teacher in-person or picked	
	up at school at the	
	beginning of the week for	
	remote learners.	
	With the bag of ingredients	
	given, how many s'mores	
	can you make? Why? (This will be done in a "Wait	
	Questions" format where	
	all students wait for 3,2,1	
	count down.)	
	 Do a few more real-life 	
	practice problems -	
	identifying the limiting and	
	excess reagents.	
	Limiting & Excess reagents	
	presentation: 25 minutes	
	Teacher presents a	
	powerpoint presentation	
	about limiting and excess	
	reagents by sharing their	
	screen. • Presentation includes	
	practice problems with	
	stoichiometric conversions	
	as well as active learning	
	opportunities throughout	
	using Zoom features (e.g.,	
	beautiful at the second at a Aug.	Ī

Small Group Break-outs:

- Heterogeneous
- Homogeneous
- Choice-based
- Project-based
- Independent work time

In heterogeneous breakout out groups (predetermined by the teacher), students first work through scaffolded practice problems and then complete 5 additional practice problems (with one participant presenting one at a time to the group) via "Practice Makes Perfect").

hand raising, chat, etc.).

30 minutes

Check for Understanding:

- Completed task
- Quick-check in
- Group/individual sharing
- Poll
- Assessment

Whole group share: 5 minutes

(each group has one person share)

- What is one thing you found challenging when working through these types of problems?
- How did you work as a group to address those challenges?
- What questions do you still have?

Check for understanding: 10 minutes

 Put up a slide with one question for everyone to complete via screen share.

If 4.95 g of ethylene (C_2H_4) are combusted with 3.25 g of oxygen. Hint:

- a. What is the limiting reagent?
- b. How many grams of CO₂ are formed?
 - All students should complete the problem independently and give a thumbs up when they are done by using the reactions feature. Once everyone is done the teacher will implement "Wait Questions" (e.g., count down 3,2,1 and then have everyone flood the chat with their answers).
 - Students should take 2 minutes to read through all of the answers and decide if they think they were correct, incorrect, etc.
 (During that time the teacher is able to get a read of the class as well.)

Clarification: 5 minutes

Based on students

20 minutes

Closing:

- Asynchronous learning opportunities & collaboration
- Next steps
- Exit ticket

Review homework:

Go through asynchronous activities for students as well as homework/ prep work to be completed prior to tomorrow's lesson.

"Virtual Non-verbal Cues"

Students are asked to put up a reaction at the end of class to show how comfortable they are with identifying limiting and excess reagents. This will help to identify small groups for tomorrow's lesson.



👏 = feeling good



👍 = needs a little more practice



= help please

5 minutes